CHARACTERIZATION AND MECHANICAL PROPERTIES OF CONCRETE MIXTURES MADE WITH SEDIMENTARY LIME AND INDUSTRIAL INCINERATOR ASH

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Abstract: This paper is a review of the research work done in order to promote industrial waste into concrete for common use in building construction, with the aim of emphasizing the feasibility of use incinerator ash (IA) as fine aggregate replacement and sedimentary lime (SL) as normal cement substantial have been investigated and compared with control samples. IA were used to replace 5%, 10%,15% up to 80% of the total mineral fine aggregate's volume and SL was used to replace 10%, 15% up to 50% of the total normal cement volume in concrete. The mixtures tests included compressive strength, tensile strength, modulus of elasticity, chloride penetration and drying shrinkage. The fresh concrete mixtures exhibited lower unit weight and acceptable workability compared to plain concrete. It was found that at the age of 28 days, the compressive strengths of concrete mixtures to decrease below the plain mixtures. In F50L25 mixtures a good mechanical properties were achieved after 28 days of curing. Given the experimental data show new possibilities for this waste materials reuse as structural concrete and give great advantages in waste minimization as well as resources conservation and reduce the cost of materials.

Keywords: Sedimentary lime, Incinerator ash, Structural concrete, cement, fine aggregate, green concrete

1.0 Introduction

Regulations on waste elimination make waste management significant problem for the public authorities today (Aubert *et al.*, 2004). So that during recent years there has been a growing emphasis on the utilization of waste Materials and by-products in construction materials (Rafat, 2010). Cement is among the most energy-intensive materials used in the construction industry and a major contributor to CO^2 in the atmosphere (Altwair and Kabir, 2010). Cement conservation is the first step in reducing the energy consumption and greenhouse-gas emissions. Resource productivity consideration will require us to minimize Portland cement use while meeting the future

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demands for more concrete (Malhotra, 1999; Mehta, 1999). Mehta (2009) said that using less cement in concrete productions can save 50% of clinker production.

Municipal solid waste ash (MSWA) is the by-product produced during the combustion of municipal solid waste. Two types of the ashes are produced as a result of the incineration process; bottom ash and fly ash. Municipal solid waste combustor ash is a relatively lightweight material compared with natural sands and aggregate (Malhotra, 2000; Chimenos *et al.*, 1999). Ferreira *et al.* (2003) published various uses of MSW fly ash, which included: (i) construction materials; (ii) geotechnical applications; (iii) agriculture; and (iv) miscellaneous applications.

Hamernik and Frantz (1991) shows that MSW fly ash could be used as partial replacement for cement in concrete mixes as a supplementary cementitious material as it contains some quantities of typical cement minerals, although in lesser quantity than cement clinker. In other research, Bertolini *et al.* (2004) showed that bottom ashes from MSW incineration are potentially attractive as mineral additions for the production of concrete. When MSWI bottom ashes were added to the concrete mix after being dry ground, strength and durability of concrete were negatively affected by entrapment of gas bubbles.

Mangialardi (2001) studied the sintering process of MSW fly ash in order to manufacture sintered products for reuse as concrete aggregates. Four types of fly ash resulting from different Italian MSW incineration plants were studied. He concluded that: sintering process of untreated MSW fly ashes proved to be ineffective for manufacturing sintered products for reuse as a construction material.

MSW fly ash can also be used in concrete as aggregate. MSW fly ash could be processed into pellets and used as lightweight aggregates Rebeiz and Mielich (1995) have advocated the possible use of MSW ash in construction applications including the innovative use in the production of polymer concrete. Pera *et al.* (1996) examined the use of MSWI bottom ash as alternative aggregate for concrete having a characteristic 28-day compressive strength of 25 MPa. They concluded that concrete with 50% bottom ash satisfy the requirements to be used as a building concrete presenting characteristic 28-day strength of 25 MPa. When the gravel is entirely replaced by bottom ash the strength decreases due to the softness of aggregates. Collivignarelli and Sorlini (2002) assessed the feasibility of concrete production using stabilized MSW incineration fly ashes in addition to natural aggregates. They concluded that MSWI fly ashes showed good chemical and physical quality for the production of concrete mixtures.

This paper summarizes the results of a research aimed at studying the use of sedimentary lime as cement substantial and industrial incinerator ash as replacement of fine aggregates for the production of structural concrete.

2.0 Materials and Methods

The study was designed in three steps. First, the binding abilities of sedimentary lime (SL) were investigated and an acceptable level of SL in total binder was determined. In the second step, incinerator ashes (IA) were use as replacement of fine aggregates and acceptable level of IA in total fine aggregates volume was determined. And at last concretes with the formerly determined level of SL and different portions of incinerator ashes were studied for their consistency, fresh and hardened concrete properties. In order to investigate the mechanical properties of concrete that containing lime as binder and incinerator ash, specimens of a cube shape of $150 \times 150 \times 150$ mm were fabricated.

2.1. Materials

2.1.1 Aggregates

Coarse aggregate: The coarse aggregate was a 25 mm and Unit weight of 1684.7 kg/m³ supplied from Charm Shahr in Tabriz, which was used together with coarse and fine sand of the same source.

Fine aggregate: The fine aggregate was natural sand of 4.75 mm maximum size obtained from Charm Shahr in the area of Tabriz in Iran. The properties of the fine aggregate was determined and fulfilled according to I.Q.S. No. 45/1984, I.Q.S. No. 30/1984 and I.Q.S. No. 31/1981.

2.1.2 Incinerator Ash

Ash particles provided by Tabriz Petrochemical Company in Iran. The Chemical composition of Incinerator Ash is presented in table 1.

| ruble 1. Chemiear composition of memerator Ash | | | | | |
|--|--------------------------------|----------|--|--|--|
| Compounds | Abbreviation | % Weight | | | |
| Iron oxide | Fe ₂ O ₃ | 47.3 | | | |
| Lime | CaO | 18.6 | | | |
| Phosphorus pent oxide | P_2O_5 | 13.5 | | | |
| Alumina | Al ₂ O ₃ | 2.27 | | | |
| Magnesia | MgO | 1.84 | | | |

 Table 1: Chemical composition of Incinerator Ash

Incinerator bottom ash is the coarse residue left on the grate of waste incinerators. Incineration is a waste treatment process that involves the combustion of organic substances contained in waste materials (oil and biological sludge). Combustion of these materials creates solid wastes called ash, which can contain any of the elements that were originally present in the waste. Incinerators reduce the solid mass of the original waste by 80–85% and the volume by 95-96 %, depending on composition and degree of

recovery of materials such as metals from the ash for recycling. Incineration markedly reduces the amount of waste for disposal and thus decreases the burden on landfill sites (Olie et al., 1977). These particles were industrial aggregate of 4.75 mm maximum size and Unit weight of 1684.7 kg/m3. Data regarding the properties of the aggregates and the ash particles are given in Table 2 and the gradation of sand and Incinerator Ash is presented in Fig. 1.

| Aggregate type | Specific gravity | Absorption (%) | Fineness modulus | Unit weight (kg/m3) |
|---------------------------|---------------------|-------------------|------------------|------------------------|
| Coarse aggregate | 2.46 | 2.56 | NA | 1684.7 |
| Fine aggregate | 2.54 | 4.91 | 3.16 | 1760.4 |
| Incinerator Ash Particles | 1.17 | 17 | 3.07 | 1160.2 |

| Table 2: Pro | perfies of aggregation | 3 |
|--------------|------------------------|---|
| | | |



Figure 1: Gradation of sand and Incinerator Ash.

2.1.3 Cement

Type I Portland cement (Soufian cement factory) was used in all types of aggregate content mixtures.

2.1.4 Lime

Sedimentary lime obtained by water refining process of Zarineh Roud river that passes from Tabriz Petrochemical region and created in two roles:

a) Precipitate calcium bicarbonates in the form of calcium carbonate according to the following chemical reactions:

$$Ca (HCO_3)_2 + Ca (OH)_2 \rightarrow 2CaCO_3 + 2H_2O$$
(1)

Magnesium bicarbonates are changed into soluble magnesium carbonates up to 50 ppm of CaCO₃ according to the following chemical reaction:

$$Mg (HCO_3)_2 + Ca (OH)_2 \rightarrow MgCO_3 + CaCO_3 + 2H_2O$$
(2)

b) By injection of an excess dose of lime, magnesium is precipitated in the form of magnesium hydroxide according to the following chemical reactions:

$$MgCO_{3} + Ca (OH)_{2} \rightarrow Mg (OH)_{2} + CaCO_{3} + 2H_{2}O$$
(3)

$$Mg^{2+} + Ca (OH)_{2} \rightarrow Mg (OH)_{2} + Ca^{2+} + 2H_{2}O$$
(4)

As magnesium hydroxide precipitates, the silica and practically all the colloids are also precipitated by adsorption. Thus ensuring a fouling index within the permissible limits for a good working of the reverse osmosis plant. A Sample of the Incinerator Ash and sedimentary lime is shown in Fig 2.



Figure 2: (a, b) Samples of waste materials: (a) Incinerator Ash, (b) sedimentary lime

2.2. Mixture Proportioning

A concrete mix, commonly used in the locality as structural concrete of 40 MPa strength, was chosen for the field trial concrete. This was used as the reference mix and other mix proportions were formulated based on this mix to include the lime as substantial of cement and incineration ash as fine aggregate replacement mentioned above. The purpose of the trial was to determine the following effects on field concrete.

The experimental setup and specimen fabrication are summarized in Tables 3 and 4, respectively. To unify the Ash and lime content, a designated percentage for each mix types were converted to a total fine aggregate and cement volume percentage. The equivalent values of Ash and lime contents are given in Table 2. Specimens were remolded 24 hour after casting, and were kept in a curing room at a temperature of 25 °C, with a relative humidity of 60%, until the time of testing. A normal, non-airentrained, Portland cement concrete, was designed as the control mix according to B.S.1881, part 7(British Standard Institution, 1952). The mix required a 0.48 watercement ratio. Other constituents are given in Table 3 for P, F, FL and LC specimens. P control mix was used as the basis for preparing of concrete mixes specified by F, FL, and LC mixes. In the F mixes, the fine aggregates of the control mix were replaced by IA, and in the FL mixes, the sand in the control mix was replaced by IA and SL were replaced by normal cement. In the LC mixes, sedimentary lime was used as replacements for cement, respectively. For a F50L25 mix, incineration ash replaced 50% of the sand volume and sedimentary lime replaced 25% of the cement volume in concrete mixtures.

| Specimen designation | Fine mineral | Coarse mineral | Cement (%) | Lime (%) | Incinerator Ash | Replication compressive |
|-------------------------|-----------------|-------------------|---------------|-------------|--------------------|----------------------------|
| _ | Aggregate | Aggregate | | | aggregate | test |
| | (%) | (%) | | | (%) | |
| Р | 100 | 100 | 100 | 0 | 0 | 3 |
| F5 | 95 | 100 | 100 | 0 | 5 | 3 |
| F10 | 90 | 100 | 100 | 0 | 10 | 3 |
| F15 | 85 | 100 | 100 | 0 | 15 | 3 |
| F20 | 80 | 100 | 100 | 0 | 20 | 3 |
| F25 | 75 | 100 | 100 | 0 | 25 | 3 |
| F30 | 70 | 100 | 100 | 0 | 30 | 3 |
| F35 | 65 | 100 | 100 | 0 | 35 | 3 |
| F40 | 60 | 100 | 100 | 0 | 40 | 3 |
| F45 | 55 | 100 | 100 | 0 | 45 | 3 |
| F50 | 50 | 100 | 100 | 0 | 50 | 3 |
| F55 | 45 | 100 | 100 | 0 | 55 | 3 |
| F60 | 40 | 100 | 100 | 0 | 60 | 3 |
| F65 | 35 | 100 | 100 | 0 | 65 | 3 |
| F70 | 30 | 100 | 100 | 0 | 70 | 3 |
| F75 | 25 | 100 | 100 | 0 | 75 | 3 |
| F80 | 20 | 100 | 100 | 0 | 80 | 3 |
| L10C90 | 100 | 100 | 90 | 10 | 0 | 3 |
| L15C85 | 100 | 100 | 85 | 15 | 0 | 3 |
| L20C80 | 100 | 100 | 80 | 20 | 0 | 3 |
| L25C75 | 100 | 100 | 75 | 25 | 0 | 3 |
| L30C70 | 100 | 100 | 70 | 30 | 0 | 3 |
| L40C60 | 100 | 100 | 60 | 40 | 0 | 3 |
| L50C50 | 100 | 100 | 50 | 50 | 0 | 3 |
| L75C25 | 100 | 100 | 25 | 75 | 0 | 3 |
| F50L25 | 50 | 100 | 75 | 25 | 50 | 3 |

Table 3: Experimental program

| Specimen | Cement (kg) | Lime (kg) | Gravel (kg) | Sand (kg) | Ash Aggregate (kg) | W/C |
|----------|-------------|-----------|-------------|-----------|--------------------|------|
| Р | 406 | 0 | 1027 | 747 | 0 | 0.48 |
| F5 | 406 | 0 | 1027 | 709.6 | 37.4 | 0.48 |
| F10 | 406 | 0 | 1027 | 672.3 | 74.7 | 0.48 |
| F15 | 406 | 0 | 1027 | 634.9 | 112.1 | 0.48 |
| F20 | 406 | 0 | 1027 | 597.6 | 149.4 | 0.48 |
| F25 | 406 | 0 | 1027 | 560.2 | 186.8 | 0.48 |
| F30 | 406 | 0 | 1027 | 522.9 | 224.1 | 0.48 |
| F35 | 406 | 0 | 1027 | 485.5 | 261.5 | 0.48 |
| F40 | 406 | 0 | 1027 | 448.2 | 298.8 | 0.48 |
| F45 | 406 | 0 | 1027 | 410.8 | 336.2 | 0.48 |
| F50 | 406 | 0 | 1027 | 373.5 | 373.5 | 0.48 |
| F55 | 406 | 0 | 1027 | 336.1 | 410.9 | 0.48 |
| F60 | 406 | 0 | 1027 | 298.8 | 448.2 | 0.48 |
| F65 | 406 | 0 | 1027 | 261.4 | 485.6 | 0.48 |
| F70 | 406 | 0 | 1027 | 224.1 | 522.9 | 0.48 |
| F75 | 406 | 0 | 1027 | 186.7 | 560.3 | 0.48 |
| F80 | 406 | 0 | 1027 | 149.4 | 597.6 | 0.48 |
| L10C90 | 365.4 | 40.6 | 1027 | 747 | 0 | 0.48 |
| L15C85 | 345.1 | 60.9 | 1027 | 747 | 0 | 0.48 |
| L20C80 | 324.8 | 81.2 | 1027 | 747 | 0 | 0.48 |
| L25C75 | 304.5 | 101.5 | 1027 | 747 | 0 | 0.48 |
| L30C70 | 284.2 | 121.8 | 1027 | 747 | 0 | 0.48 |
| L40C60 | 243.6 | 162.4 | 1027 | 747 | 0 | 0.48 |
| L50C50 | 203 | 203 | 1027 | 747 | 0 | 0.48 |
| L75C25 | 101.5 | 304.5 | 1027 | 747 | 0 | 0.48 |
| F50L25 | 50 | 100 | 1027 | 373.5 | 373.5 | 0.48 |

Table 4: Concrete mixture proportions

2.3 Specimen and Tests of Specimens

Cubes of concrete of $150 \times 150 \times 150$ mm were molded for compressive strength, and fresh and dry density tests.

2.3.1 Test of Specimens

- 1. Casting, compaction and curing: Accomplished according to B.S.1881, part 7 and B.S.1881, part 6 (British Standard Institution, 1952).
- 2. Slump test: Fulfilled according to B.S.1881, part 2.
- 3. Fresh densities: Measured for all cubes after molding and compacting immediately according to B.S.1881, part 5. The fresh density represents the mean of fresh densities for 3 cubes.

- 4. Compression strength test: Concrete cubes were prepared according to B.S.1881, part 7. The Forney machine was used for the compression test. The cubes were tested immediately after taken out of water while they were still wet. The average of compression strength of 3 cubes was recorded for each testing age.
- 5. Tensile strength and modulus of elasticity: The tensile splitting strength is carried out according to ASTM C 496. The static modulus of elasticity indices were implemented according to ASTM C469.
- 6. Chloride Penetration and Expansion Characteristics: The drying shrinkage test is carried out according to ASTM C157 and rapid chloride permeability test (RCLP), using the procedure described in ASTM C-1202.

3.0 Results and Discussion

3.1 Properties of Fresh Concrete

The properties of the fresh concrete including the slump, Water absorption and unit weight are as follows.

3.1.1 Slump Test

The results of the slump tests indicate that the slump is decreasing with increasing the lime and ash particles ratio. The reductions of slump are 39.29%, 67.72%, 77.54%, 88.42%, 30.17%, 42.57%, 46.66% and 64.31% for F20, F40, F60, F80, L25C75, L50C50, L75C25 and F50L25, respectively. This reduction may be attributed to the fact that the water absorption of lime and ash particles is more than the normal cement and natural fine aggregate. So that the replacement of lime and incinerator ash in concrete mixtures reduced workability and in order to have normal concrete workability, the mixture needs extra water. In precast applications and large sites based on the following consideration: Workability has a broad range from very low (at slump = 0–25 mm) applied for vibrated concrete in roads or other large sections, to high workability (at slump = 100–180 mm) applied for sections with congested reinforcement (Koehler and Fowler, 2003). Concrete products with lime as binder and incinerator ash are able to use in low slump needs in ordinary constructions. The replacement of lime and incinerator ash in concrete ash in the same time in concrete mix decreasing the workability more than other components. In LC mixtures the slump reduction is lower than F mixtures.

3.1.2 Water Absorption

Water absorption tests were carried out on cube specimens. The results indicate that concretes mixtures containing lime and ash particles showed higher water absorption ratios than conventional concretes. But the water absorption of the F10, F20 and F50 is lower than conventional concretes and F60 has similar absorption in caparison with plain concrete. The water absorption of the F30, F40, F70, F80, L25C75, L50C50, L75C25 and F50L25 mixes were increased 9.93%, 12.26%, 7.48%, 11.11%, 26.09%, 37.33% 40.61% and 4.22% compared to plain concrete. But in the F10, F20 and F50 mixes the water absorption were decreased 4.41%, 1.48%, 4.41% respectively. The porosity of the adhered mortar causes the water to penetrate into the accessible pores and leads to an increase in the water absorption capacity in comparison with conventional concrete.

3.1.3 Unit Weight

The unit weight of the concrete ranged from 2386 to 2156 kg/m3, depending on ash and lime contents. The results (Table 5 and Fig. 3) indicate that the Unit weight tends to decrease by different percentage of lime and ash particles replacement in normal concrete. This trend may be attributed to the density of the lime and ash particles being lower than the normal cement and sand by 22.9% and 34.1%, which leads to a reduction in the density. Thus, concrete containing incinerator ash and lime could be used wherever lightweight concrete is required. For example, this kind of concrete can be used in structures to reduce earthquake damage. The F80 mix that incineration ash replaced 80% of the sand volume has lowest unite weight. Therefore, with increasing amount of lime and ash particles volume in concrete the unit weight will decrease.



Figure 3: Unit weight of concrete mixtures

| Specimen | Unit weight (Kg/m ³) | Specimen | Unit weight (Kg/m ³) |
|----------|----------------------------------|----------|----------------------------------|
| Р | 2386 | F70 | 2175 |
| F5 | 2352 | F75 | 2164 |
| F10 | 2340 | F80 | 2156 |
| F15 | 2337 | L10C90 | 2346 |
| F20 | 2331 | L15C85 | 2311 |
| F25 | 2301 | L20C80 | 2293 |
| F30 | 2297 | L25C75 | 2271 |
| F35 | 2267 | L30C70 | 2269 |
| F40 | 2245 | L40C60 | 2267 |
| F45 | 2240 | L50C50 | 2265 |
| F50 | 2232 | L75C25 | 2254 |
| F55 | 2225 | F50L25 | 2227 |
| F60 | 2210 | | |

Table 5: Unit weight of concrete containing waste industrial material

3.2 Hardened Concrete Properties

3.2.1 Compressive Strength

The compressive strength test results for the concrete mixtures are presented in Fig. 4 and loading states of cubic specimens containing lime are shown in Fig. 5. The results show a tendency for compressive strength values of concrete mixtures to decrease below the plain mixtures.

The L75C25 mixes weren't suitable among the other samples. The samples that lime completely replaced by cement, the cubic specimen was decayed after bring up from water. The reduction of compressive strength in mixtures containing ash particles can be attributed to the decrease in adhesive strength between the ash particles and the cement paste. The higher water absorption of ash particles may restrict the water necessary for cement hydration from entering through the structure of the concrete specimens during the curing period. The Hardness of Ash particles is lower than the natural sand particles so it may be attributed to the compressive strength of the concrete containing ash particles being lower than the concrete containing sand as fine aggregate. Adhesive strength in the mixtures that lime replaced with cement is lower than cement and it may cause to decrease the compressive strength. Except the L50C50 and L75C255 mixes, all of the compressive strength values are higher than the minimum compressive strength required for structural concrete which is 17.24 MPa. The research shows that the best volume of replacing lime with normal cement in order to production of structural concretes is less than 25 percent of cement volume in concrete mixtures. In case that we want to use lime and incineration ash in same mixture the suggested mixture is F50L25.



Figure 4: Compressive strength of concrete containing waste industrial material (7 days and 28 days).



Figure 5: (a, b, c) Testing compressive strength of concrete specimens: (a) specimens before testing, (b) Loading of specimens, (c) after testing.

3.2.2 Tensile Strength

A comparative study of the tensile splitting strength provides the results summarized in Table 6 and Fig 6, which shows the average values and variations obtained for each concrete in comparison with P mixes. These results show that the tensile splitting strength of concrete mixtures is prone to decrease with the increase of the ash particles and lime ratio in these mixtures. This trend can be attributed to the decrease in adhesive strength between the surface of ash particles and the cement paste which may limit the hydration of cement. Therefore the hydration developed slightly with time.

| Specimen | f _{ct.28} (MPa) | Specimen | f _{ct.28} (MPa) |
|----------|--------------------------|----------|--------------------------|
| Р | 3.80 | F70 | 2.66 |
| F5 | 3.80 | F75 | 2.65 |
| F10 | 3.78 | F80 | 2.59 |
| F15 | 3.63 | L10C90 | 3.61 |
| F20 | 3.58 | L15C85 | 3.55 |
| F25 | 3. 27 | L20C80 | 3.33 |
| F30 | 3.15 | L25C75 | 3.25 |
| F35 | 3.08 | L30C70 | 3.17 |
| F40 | 3.03 | L40C60 | 3.09 |
| F45 | 3.00 | L50C50 | 2.40 |
| F50 | 2.90 | L75C25 | 1.31 |
| F55 | 2.84 | F50L25 | 3.7 |
| F60 | 2.76 | | |

Table 6: Average tensile strength for the test age of 28 days



Figure 6: Influence of IA and SL content on the concrete tensile strength.

3.2.3 Modulus of Elasticity

The results of the static modulus of elasticity at different ages are given in Table 7, which includes the average Values and variations obtained for each material in comparison with P mixes. In this way, specimens with the lowest static modulus of elasticity were those corresponding to concrete containing lime as cement substitute. The modulus of elasticity for samples has been reported to be in the range of 30-96% the normal concrete.

| Specimen | E _{c.28} (MPa) | Specimen | E _{c.28} (MPa) |
|----------|-------------------------|----------|-------------------------|
| P | 31958.32 | F70 | 19360.96 |
| F5 | 31722.12 | F75 | 19123.84 |
| F10 | 30714.38 | F80 | 18571.41 |
| F15 | 29750.24 | L10C90 | 31142.64 |
| F20 | 28940.87 | L15C85 | 28588.60 |
| F25 | 26547.93 | L20C80 | 26728.49 |
| F30 | 24887.74 | L25C75 | 24901.32 |
| F35 | 24219.45 | L30C70 | 21640.42 |
| F40 | 23059.65 | L40C60 | 18772.29 |
| F45 | 22717.63 | L50C50 | 18253.57 |
| F50 | 21775.19 | L75C25 | 10041.29 |
| F55 | 20974.43 | F50L25 | 20863.34 |
| F60 | 20527.67 | | |

Table 7: Average static modulus of elasticity for the test age of 28 days

3.2.4 Shrinkage Characteristics

Volume changes in concrete due to hydration reactions, drying, wetting and drying and thermal variation occur even without application of any loads. Under normal field conditions, drying shrinkage is an important phenomenon which induces time dependent deformations and is sometimes called creep under zero loads. The results of the drying shrinkage at different ages are given in Table 8. Fig 7 presents the test results for the shrinkage of concrete with 10, 20, 30, 40 and 50% IA. The results show that the shrinkage of concrete containing incinerator ash was lower than that of concrete without incinerator ash, but concretes mixtures containing lime showed higher shrinkage than conventional concretes. In F50L25 mixes the shrinkage of concrete is similar to P mixes.



Figure 7: Test results of drying shrinkage of concrete containing IA (F10, F20, F30, F40 and F50).

| Specimen | 3 Day | 7 Day | 28 Day | 60 Day | 80 Day | 90 Day |
|----------|-------|-------|--------|--------|--------|--------|
| P | 220 | 335 | 660 | 732 | 735 | 739 |
| F5 | 218 | 333 | 651 | 730 | 734 | 737 |
| F10 | 217 | 331 | 648 | 728 | 731 | 735 |
| F15 | 215 | 329 | 633 | 720 | 728 | 729 |
| F20 | 212 | 328 | 620 | 652 | 701 | 711 |
| F25 | 191 | 318 | 587 | 612 | 665 | 685 |
| F30 | 179 | 301 | 574 | 590 | 621 | 641 |
| F35 | 154 | 188 | 531 | 562 | 598 | 611 |
| F40 | 131 | 172 | 495 | 510 | 525 | 586 |
| F45 | 111 | 159 | 471 | 461 | 477 | 578 |
| F50 | 98 | 133 | 422 | 438 | 449 | 493 |
| F55 | 87 | 119 | 387 | 411 | 420 | 461 |
| F60 | 72 | 99 | 356 | 309 | 357 | 400 |
| F65 | 60 | 87 | 321 | 288 | 311 | 352 |
| F70 | 56 | 72 | 205 | 255 | 286 | 318 |
| F75 | 48 | 64 | 188 | 241 | 269 | 263 |
| F80 | 32 | 42 | 150 | 222 | 231 | 238 |
| L10C90 | 213 | 350 | 659 | 669 | 729 | 738 |
| L15C85 | 215 | 349 | 661 | 675 | 732 | 739 |
| L20C80 | 221 | 341 | 666 | 687 | 738 | 744 |
| L25C75 | 228 | 336 | 669 | 733 | 738 | 749 |
| L30C70 | 228 | 338 | 673 | 739 | 741 | 755 |
| L40C60 | 229 | 340 | 680 | 744 | 747 | 758 |
| L50C50 | 230 | 340 | 687 | 749 | 752 | 760 |
| L75C25 | 236 | 342 | 701 | 749 | 767 | 767 |
| F50L25 | 219 | 331 | 663 | 732 | 733 | 736 |

Table 8: Drying Shrinkage (micro-strain) Data (3, 7, 28, 60, 80, 90 Day)

3.2.5 Chloride Resistance

This test method consists of monitoring the amount of electrical current passed through 51-mm thick slices of 102-mm nominal diameter cores or cylinders during a 6-h period. A potential difference of 60 V dc is maintained across the ends of the specimen, one of which is immersed in a sodium chloride solution, the other in a sodium hydroxide solution. The chloride ion penetration of concrete mixtures was determined at 28 and 90 days of age. Table 9 and 10 summarizes the rapid chloride penetrability data at 28 day and 90 day, respectively. Based on the results, concrete with IA and SL together showed the lowest penetration of chloride ions. This behavior can be related to pore refinement and decrease in permeability produced by pozzolanic reaction.

| Specimen | W / (C+PP) | Charge Passed (coulombs) | Chloride Penetrability |
|----------|------------|--------------------------|------------------------|
| F5 | 0.4 | 3172 | Moderate |
| F10 | 0.4 | 3018 | Moderate |
| F15 | 0.4 | 2911 | Moderate |
| F20 | 0.4 | 2838 | Moderate |
| F25 | 0.4 | 2745 | Moderate |
| F30 | 0.4 | 2665 | Moderate |
| F35 | 0.4 | 2574 | Moderate |
| F40 | 0.4 | 2466 | Moderate |
| F45 | 0.4 | 2315 | Moderate |
| F50 | 0.4 | 2271 | Moderate |
| F55 | 0.4 | 2241 | Moderate |
| F60 | 0.4 | 2199 | Moderate |
| F65 | 0.4 | 2175 | Moderate |
| F70 | 0.4 | 2151 | Moderate |
| F75 | 0.4 | 2133 | Moderate |
| F80 | 0.4 | 3120 | Moderate |
| L10C90 | 0.4 | 3150 | Moderate |
| L15C85 | 0.4 | 3113 | Moderate |
| L20C80 | 0.4 | 3076 | Moderate |
| L25C75 | 0.4 | 3012 | Moderate |
| L30C70 | 0.4 | 2975 | Moderate |
| L40C60 | 0.4 | 2942 | Moderate |
| L50C50 | 0.4 | 2915 | Moderate |
| L75C25 | 0.4 | 2863 | Moderate |
| F50L25 | 0.4 | 2013 | Moderate |

Table 9: Rapid Chloride Penetrability Data (28 Day)

| Specimen | W / (C+PP) | Charge Passed (coulombs) | Chloride Penetrability |
|----------|------------|--------------------------|------------------------|
| F5 | 0.4 | 918 | Very Low |
| F10 | 0.4 | 818 | Very Low |
| F15 | 0.4 | 780 | Very Low |
| F20 | 0.4 | 756 | Very Low |
| F25 | 0.4 | 712 | Very Low |
| F30 | 0.4 | 696 | Very Low |
| F35 | 0.4 | 657 | Very Low |
| F40 | 0.4 | 645 | Very Low |
| F45 | 0.4 | 609 | Very Low |
| F50 | 0.4 | 587 | Very Low |
| F55 | 0.4 | 543 | Very Low |
| F60 | 0.4 | 511 | Very Low |
| F65 | 0.4 | 501 | Very Low |
| F70 | 0.4 | 487 | Very Low |
| F75 | 0.4 | 477 | Very Low |
| F80 | 0.4 | 465 | Very Low |
| L10C90 | 0.4 | 1016 | low |
| L15C85 | 0.4 | 1015 | low |
| L20C80 | 0.4 | 971 | Very Low |
| L25C75 | 0.4 | 944 | Very Low |
| L30C70 | 0.4 | 942 | Very Low |
| L40C60 | 0.4 | 872 | Very Low |
| L50C50 | 0.4 | 804 | Very Low |
| L75C25 | 0.4 | 776 | Very Low |
| F50L25 | 0.4 | 374 | Very Low |

Table 10: Rapid Chloride Penetrability Data (90 Day)

4.0 Conclusions

From the results presented in this paper, using concrete Containing Incinerated Industrial Ash and sedimentary lime, the main conclusions are:

- 1. The slump values of concrete mixtures showed a tendency to decrease below the slump of the reference concrete mixture. Workability of new concrete mixtures was reduced with increasing lime and ash concentration. According the workability of those mixtures, those mixtures are pretty good to work based on the consideration that has a low workability for different applications.
- 2. The unit weights values of concrete mixtures with increasing lime and ash particles concentrations present lower unit weights compared to plain concrete. Using presented concrete mixtures were reduced the total weight of concrete structures.
- 3. Satisfactory quality of concrete with low strength requirements (compressive strength up to 40 MPa) may be achieved using two waste materials, namely SL and IA, simultaneously. The compressive strength values of all concrete mixtures tend to

decrease below the values for the reference concrete mixtures with increasing the ash particles and lime ratio. The study indicated the possibility that IA reduces cohesion in concrete.

- 4. A reduction in the static elastic modulus of elasticity was observed in all the mixtures.
- 5. According to the results it may be recommended to replace up to 25 % of cement by SL and to use such binder where a low strength of concrete elements is required. And the proper content for IA should not exceed 50% of fine aggregate volume. The proper content of SL and IA in the FL mixes was F50L25 mixtures.
- 6. For enhancing the chloride penetration resistance, the addition of IA and SL was found to be effective. Addition of IA as fine aggregate replacement is proved to be an efficient way to improve the dry shrinkage of concrete.
- 7. Use of IA as aggregates and SL as cement replacement has two main advantages:
- (a) It creates opportunities to decrease carbon emissions and reduces the use of new virgin aggregate and the associated environmental costs of exploitation and transportation and
- (b) It reduces unnecessary landfill of valuable materials that can be recovered. These results demonstrated the potential of reusing an abundant industrial by-product, at present Land filled, for the production of Structural concretes.

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References

- Altwair, N.M., and Kabir, S.H. (2010). Green concrete structures by replacing cement with pozzolanic materials to reduce greenhouse gas emissions for sustainable environment. 6th International Engineering and Construction Conference. Cairo, Egypt, June, 28-30.
- Aubert, J.E., Husson, B., and Vaquier, A. (2004). Use of municipal solid waste incineration fly ash in concrete. *Cement and Concrete Research*. 34, 957–963.
- Bertolini, L., Carsana, M., Cassago, D., Curzio, A.Q., and Collepardi, M. (2004). MSWI ashes as mineral additions in concrete. *Cement and Concrete Research*, 34, 1899–1906.
- British Standard Institution (BSI) (1952). Methods of testing concrete, B.S.1881, BSI, London.

- Chimenos, J.M., Segarra, M., Fernández, M.A. and Espiell, F. (1999). Characterization of the bottom ash in a municipal solid waste incinerator. *Journal of Hazardous Materials*. A64, 211–222.
- Collivignarelli, C., and Sorlini, S. (2002). Reuse of municipal solid wastes incineration fly ashes in concrete mixtures. *Waste Management*, 22, 909–12.
- Ferreira, C., Ribeiro, A., and Ottosen, L. (2003). possible applications for municipal solid waste ash. *Journal of Hazardous Materials*. B96, 201–216.
- Hamernik, J.D., and Frantz, G.C. (1991). Physical and chemical properties of municipal solid waste fly ash. *ACI Materials Journal*, 88(3), 294–300.
- Koehler, E.P., and Fowler, D.W. (2003). Measuring the workability of high fines concrete for aggregate. *International Center for Aggregate Research*, ICAR 105, University of Texas, Austin.
- Malhotra, V.M. (1999). aking Concrete Greener with Fly Ash, *Concrete International*. V. 21, No. 5, pp. 61-66.
- Malhotra, V.M. (2000). Role of supplementary cementing materials in reducing greenhouse gas emissions, In: Gjorv, O.E., and Sakai, K. (Eds.), *Concrete technology for a sustainable development in the 21st Century*. E&FN Spon, p. 226–35, London.
- Mangialardi, T. (2001). Sintering of MSW fly ash for reuses as a concrete aggregate. *Journal of Hazardous Materials*, B87, 225–239.
- Mehta, P.K. (1999). Advancements in Concrete Technology. Concrete International. pp.69-76.
- Mehta, P. K. (2009). Global Concrete Industry Sustainability. *Concrete International*. Vol.31, No. 2, Feb, pp. 45-48.
- Olie, K., Vermeulen, P.L., and Hutzinger, O. (1977). Chlorodibenzo-p-dioxins and chlorodibenzofurans are trace components of fly ash and flue gas of some municipal incinerators in The Netherlands. *Chemosphere* 6. 455–459.
- Pera, J., Coutaz, L., Ambroise, J., and Chababbet, M. (1996). Use of Incinerator bottom ash in concrete. *Cement and Concrete Research*, 27(1), 1–5.
- Rafat, S. (2010). Use of municipal solid waste ash in concrete. *Resources, Conservation and Recycling*. 55, 83-91.
- Rebeiz, K.S., and Mielich, K.L. (1995). Construction use of municipal-solid-waste ash. *Journal* of Energy Engineering, 121 (1), 2–13.